Bohr Radius

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August 20, 2020

In today's lecture we proposed a simple way to find the Bohr radius a_0 . The main idea is to consider the energy of the electron in the frame of the inertial frame of the proton

$$E = T + V = \frac{p^2}{2m} - \frac{kq^2}{a_0}$$

and then set $\frac{\partial E}{\partial a_0} = 0$ and solve for a_0 . An intermediate consideration was the positionmomentum uncertainty principle, which helped us write $\frac{p^2}{2m} \sim \frac{\hbar^2}{2ma_0^2}$. But I think the application of the uncertainty principle is tenuous, since the uncertainty principle is really saying $\Delta a_0 \Delta p \sim \hbar$, not $a_0 p \sim \hbar$.

The variational principle offers a more rigorous calculation of the Bohr radius:

Using some trial wave function $\psi(r) = Ae^{-\alpha r}$, where $\alpha \equiv \frac{1}{a}$, we can normalize to find A:

$$1 = \iiint |\psi|^2 \,\mathrm{d}\vec{r} = 4\pi \int_0^\infty r^2 |\psi|^2 \,\mathrm{d}r$$
$$= 4\pi A^2 \int_0^\infty r^2 e^{-2\alpha r} \,\mathrm{d}r$$
$$= \frac{A^2 \pi}{\alpha^3} = 1$$
so we find $A = \sqrt{\frac{\alpha^3}{\pi}}$

Now we seek to find some α_{optimal} such that $\langle H \rangle (\alpha_{\text{optimal}})$ is a minimum. Note

$$\langle H \rangle(\alpha) = \langle V \rangle(\alpha) + \langle T \rangle(\alpha)$$

Let's find the expectation values of the potential and kinetic energies:

$$\begin{split} \langle V \rangle &= \int \! \int \! \int V(r) |\psi|^2 \, \mathrm{d}\vec{r} \\ &= -4\pi \int r^2 \frac{kq^2}{r} e^{-2\alpha r} \\ &\text{integration...} = -kq^2 \alpha \end{split}$$

Meanwhile,

$$\langle T \rangle = \iiint \psi^* \left(\frac{\hbar}{2m} \frac{\partial^2}{\partial r^2}\right) \psi \, \mathrm{d}\vec{r}$$

by parts: $= \frac{4\pi\hbar^2}{2m} \int r^2 |\frac{\partial\psi}{\partial r}|^2 \, \mathrm{d}r$
 $= \frac{4\pi\hbar^2}{2m} A^2 \int r^2 \frac{\partial e^{-2\alpha r}}{\partial r} \, \mathrm{d}r$
integration... $= \frac{\hbar\alpha^2}{2m}$

 So

$$\langle H \rangle(\alpha) = \langle V \rangle(\alpha) + \langle T \rangle(\alpha) = -kq^2\alpha + \frac{\hbar\alpha^2}{2m}$$

We want α_{optimal} such that $\langle H \rangle (\alpha_{\text{optimal}})$ is a minimum:

$$\frac{\partial \langle H \rangle}{\partial \alpha} = -kq^2 + \frac{k^2\alpha}{m} = 0$$

Solving, we find that $\alpha_{\text{optimal}} = \frac{kq^2m}{\hbar^2}$. But $\alpha \equiv \frac{1}{a}$, so the Bohr radius is

$$a_0 = \frac{\hbar}{kq^2m}$$

Incidentally,

$$\langle H \rangle (\alpha_{\text{optimal}}) = \langle V \rangle (\alpha_{\text{optimal}}) + \langle T \rangle (\alpha_{\text{optimal}}) = \frac{k^2 q^4 m}{2\hbar^2} - \frac{k^2 q^4 m}{\hbar^2} = -\frac{k^2 q^4 m}{2\hbar^2} = -R_y$$

where R_y is the Rydberg energy. It is nice that $\frac{1}{2}\langle V \rangle = \langle T \rangle$. This is what we expect from the virial theorem.